CALCIUM PHOSPHATES (CaP) OCCURRENCE AND PROPERTIES

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OCCURRENCE

PHOSPHATES

CALCIUM

This type of material is of special significance for human beings because they represent the inorganic part of major normal (bones, teeth and antlers) and pathological (i.e., those appearing due to various diseases) calcified tissues of mammals. For example, atherosclerosis, and dental caries Calcium phosphate (CaP) is the mutual name of minerals family that contains the calcium cations (Ca^{+2}) together with different ionic phosphorous



Ca/P molar ratio		Compounds and their typical abbreviations	Chemical formula	Solubility product at 25°C, - log(Ks)	Solubility at 25°C, g/L	pH stability range in aqueous solutions at 25°	°C
l	0.5	Monocalcium phosphate monohydrate (MCPM)		1.14	~ 18 🤺	0.0 - 2.0	
	0.5	Monocalcium phosphate anhydrous (MCPA)			~ 17	[c]	
	1.0	Dicalcium phosphate dihydrate (DCPD), mineral brushite	XIIII		~0.088	2.0 - 6.0	
	1.0	Dicalcium phosphate anhydrous (DCPA), mineral monetite			~0.048	[c]	
	1.33	Octacalcium phosphate (OCP)	MANNIN MARCE		~0.0081	5.5 - 7.0	
	1.5	a -Tricalcium phosphate (a-TCP)	MARINE MARCH		~0.0025	[a]	
	1.5	β -Tricalcium phosphate (β -TCP)			~0.0005	[a]	
	1.2 - 2.2	Amorphous calcium phosphates (ACP)	Ca _x H.		[b]	~ 5 - 12 [d]	
	1.5 - 1.67	Calcium-deficient hydroxyapatite (CDHA)	Ca10-		~0.0094	6.5 - 9.5	
	1.67	Hydroxyapatite (HA)			~0.0003	9.5 - 12	
	1.67	Fluorapatite (FA)	Constant and a second	0.0	~0.0002	7 - 12	
	1.67	Oxyapatite (OA)	Ca ₁₀ (~ 69	~0.087	[a]	
	2.0	Tetracalcium phosphate (TTCP)	Ca ₄ (PO ₄) ₂ O	38 - 44	~0.0007	[a]	

[a] These compounds cannot be precipitated from aqueous solutions. [b] Cannot be measured precisely. [c] Stable at temperatures above 100°C. [d] Always metastable.





CaP Characterization

techniques



Spectroscopic Techniques

- FTIR
- Raman spectra
- XPS
- XRD
- EDS

Direct visualization

- SEM or FESEM
- TEM
- AFM





FTIR

- Measures the intensity over a narrow range of wavelengths at a time.
- no external calibration is required.
- Identification of even small concentrations of contaminants.

Haider, A., Haider, S., Han, S. S. and Kang, I.-K. (2017). Recent advances in the synthesis, functionalization and biomedical applications of hydroxyapatite: a review. *Rsc Advances*, 7, 7442-7458.





FTIR

Inorganic materials are not easily analyzed by FTIR spectroscopy

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FTIR



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FTIR



peaks at 560 and 1017 cm-1 representing HPO4 and PO4, respectively. peak at 1460 cm-1 indicated CO3.





Raman spectra

- Highly specific
- inorganic materials are easier to analyze







Raman spectra

- The detection requires a sensitive and highly optimized instrument
- Fluorescence of the impurities or of the sample itself can hide the Raman Spectrum
- sample heating.





Raman spectra









Raman spectra



Garma, Noor Man and Ali I. Ibrahim. "Development of a remineralizing calcium phosphate nanoparticle-containing selfetching system concreted on ding." *Clinical Oral Investigations* 27.4 (2023): 1483-1497.





XPS

 Provides unique information about the chemical composition of a material





XPS

- Slow
- poor spatial resolution
- requires high vacuum





XPS







XRD

- Powerful and rapid (<20 min) technique
- provides an unambiguous mineral determination
- data interpretation is relatively straightforward





XRD

- homogeneous and single phase materials are best for identification
- Peak overlay may occur





XRD









The XRD pattern of nHA (blue curve) and nHA-SEP (pink curve) exhibited different sharp diffraction peaks located at 20 =10.82 (100), 21.81(200), 22.90 (111), 25.87 (002), 28.126(102), 28.96(210), 31.77(211), 32.19 (112), 32.9(300), 34.04(202), 35.48 (301), 42.02(311), 43.8(113), 45.3 (203), 46.71(222), 48.10 (312), 49.46 (213), 50.49 (321), 51.28 (410), 52.10° (402), and 53.14° (004). Garma, Noor Milland Ali I. Ibrahim. "Development of a remineralizing calcium phosphate nanoparticle-containing self-etching system for orthocol for bonding." *Clinical Oral Investigations* 27.4 (2023): 1483-1497.





EDS

- Chemical microanalysis technique.
- provides unique peaks characteristic of the atomic structure of the **atoms**.
- quick and versatile technique.





EDS

- semi-quantitative.
- Comparatively lower precision.





EDS









Techniques



FESEM, TEM, and AFM provide information about the shape, size, morphology, and distribution of the CaP nanoparticles (Haider et al., 2017; Balhuc et al., 2021)



Techniques



AFM

- High size resolution
- 3D profile.





Techniques



AFM

- Slow speed
- limited scanning area





Techniques







Techniques



TEM, SEM or FE-SEM Direct visualization, high resolution

















Techniques



FE-SEM





CONCLUSION

- In spite of almost the 250-year-long history of the CaP.research and many important discoveries, still many gaps remain in our knowledge to be investigated in future.
 - The knowledge of the specifications and limitations of different characterization techniques entail opportunities to accomplish the task of filling this gap of knowledge.

