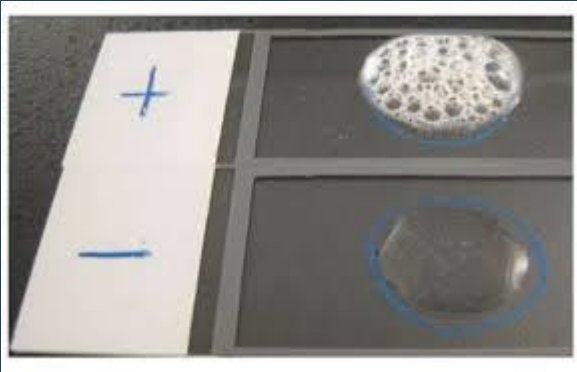


# MOLECULAR DOCKING

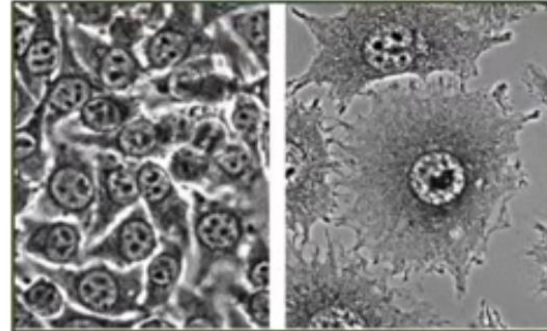
Presented By  
Dr. Rawaa Saladdin Jumaa

# Biochemical Reactions



## Catalase test

Catalase enzyme convert  
 $H_2O_2$  into  $H_2O + O_2$



## Target

Cells change morphology  
on addition of an inhibitor



## Ligand binding assay

Color of the reaction  
mixture change on ligand  
binding

# Molecular Recognition

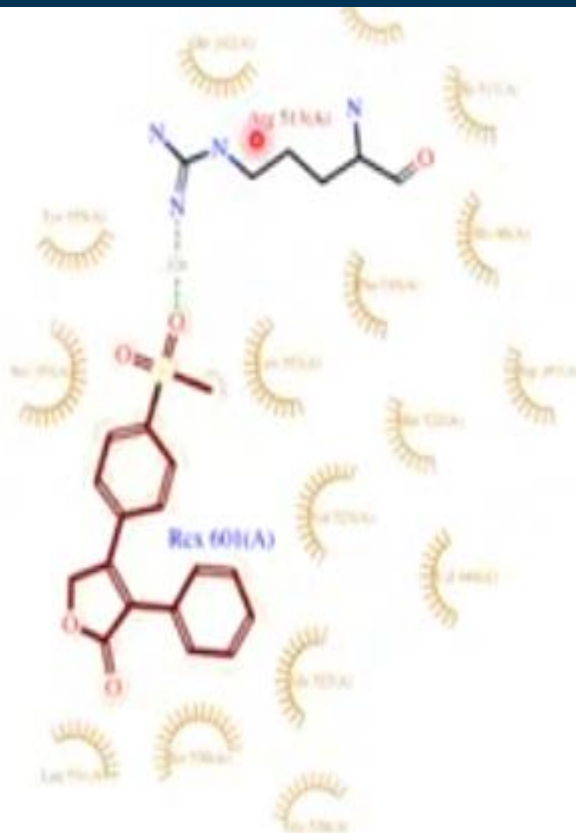
- Every cellular process happens due to interactions between molecules
- These interactions are governed by various intermolecular forces of attraction
- Goal of docking = given structure of 2 biomolecules, determine if
  - They interact favorably
  - If Yes, then what is the orientation which maximizes interaction & minimizes energy

# Molecular Interaction

## Interatomic Interaction

- Enzyme substrate complex is formed with geometric as well as electrostatic complementarity
- A large number of interaction contribute towards molecular recognition, include
  - Van der Waals
  - Dipole-Dipole
  - Ion-Dipole
  - Ion- Ion
  - Hydrogen bond (receptor ligand interaction)
  - Covalent interactions

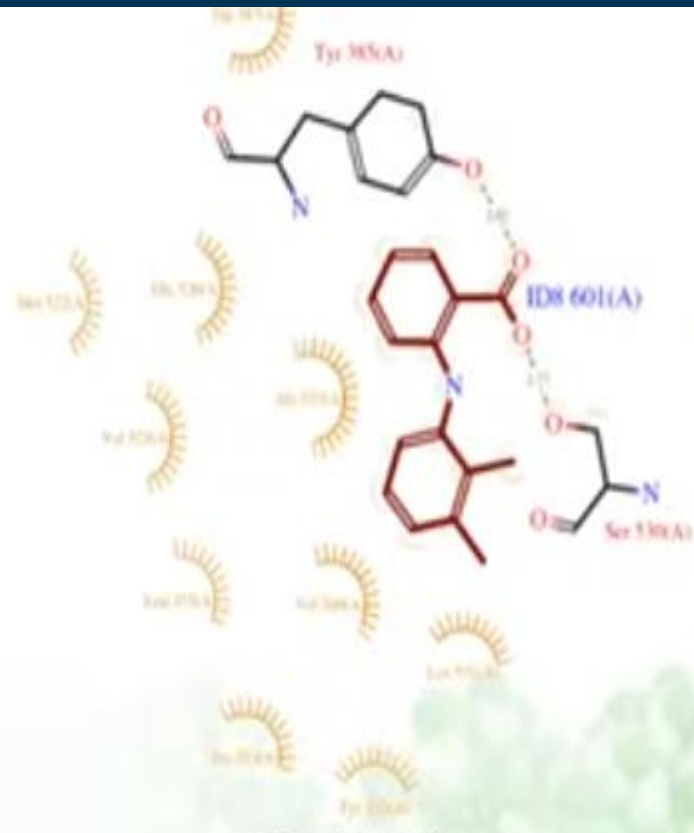
# H bond Contribution



**PDB ID: 5KIR**

Vioxx (Rofecoxib) Bound to Cox-2

IC50: 3400nM



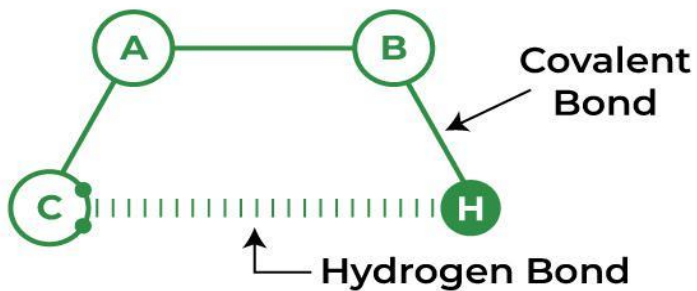
**PDB ID: 5IKR**

Mefenamic Acid Bound to Cox-2

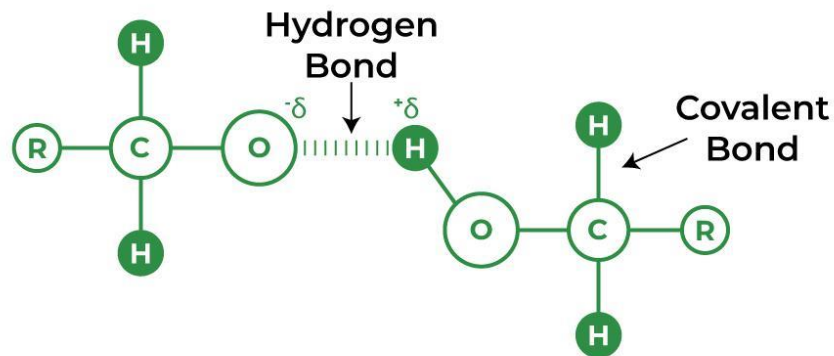
IC50: 2900nM

# H bond Contribution

## Intramolecular Hydrogen Bonding

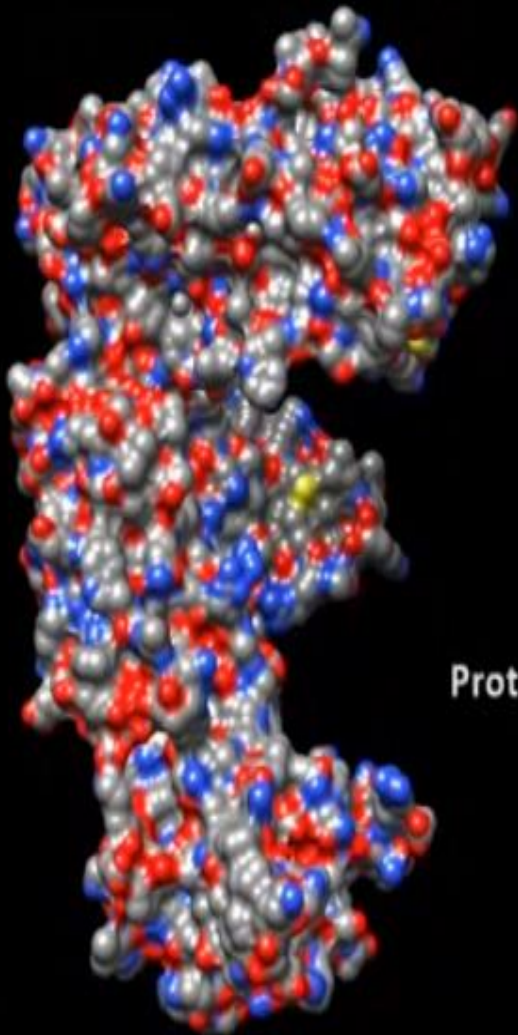


eg. :- HF, Ethylene glycol, etc.



Hydrogen Bonding in Alcohols

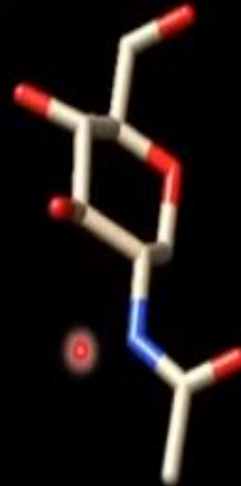
# Bio-molecular Structure



Proteins

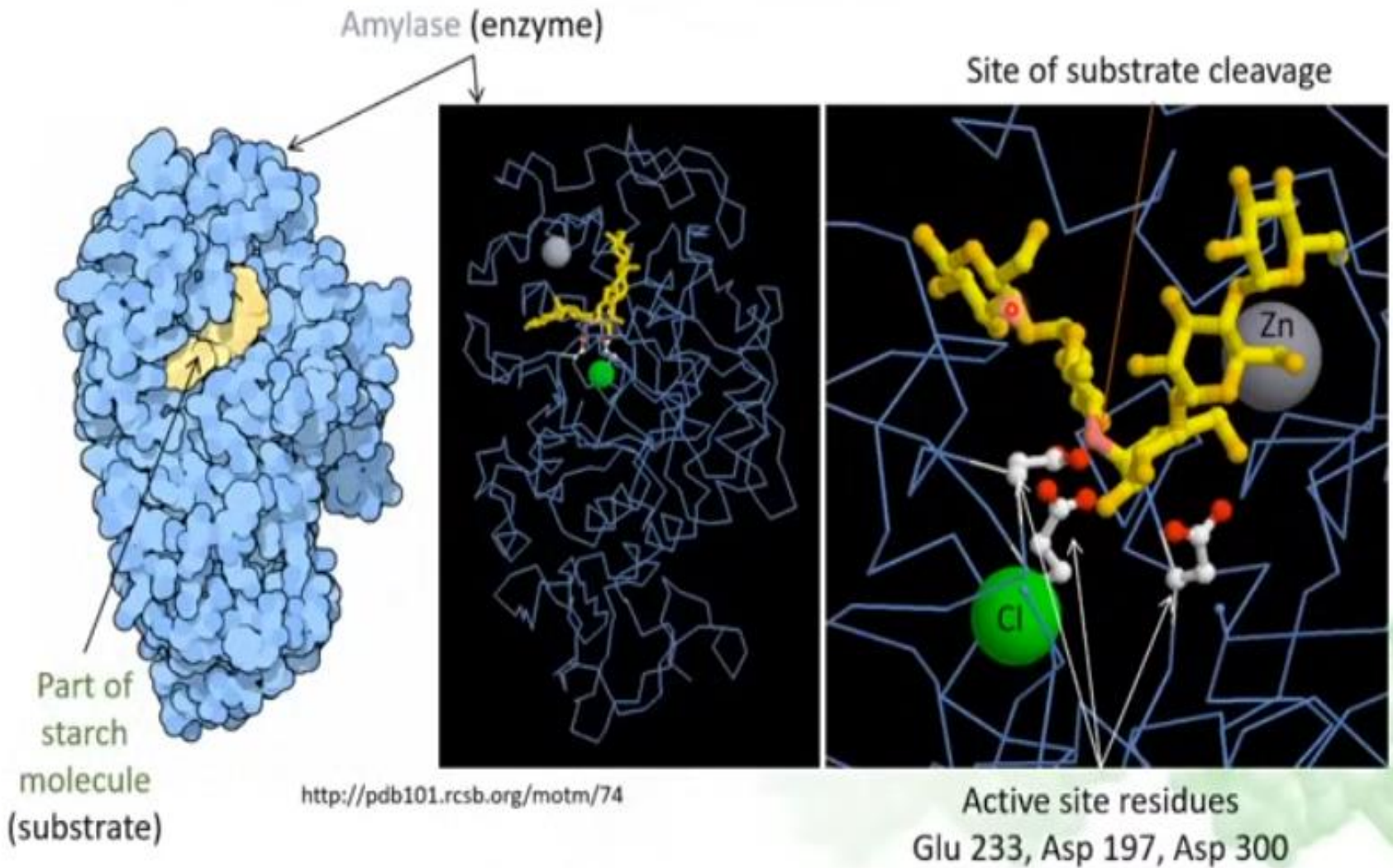


Nucleic Acids



Other like lipids,  
carbohydrates,  
sugars and many  
other molecules

# Sites



# Binding Sites

## PROTEIN STRUCTURE

Scaffold to support and position active site

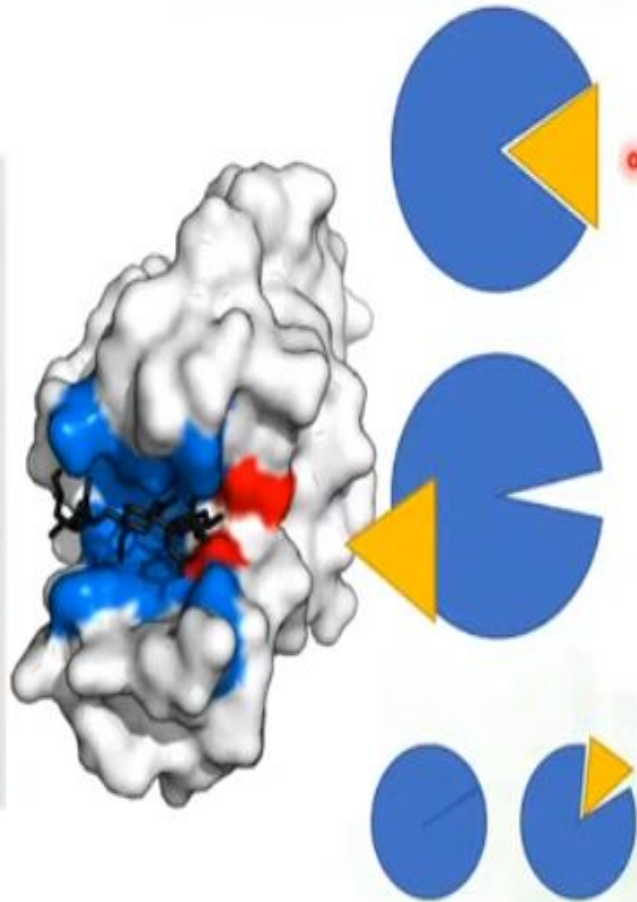
## ACTIVE SITE

### BINDING SITES

Bind and orient substrate(s)

### CATALYTIC SITE

Reduce chemical activation energy



## Binding Site

Ligand molecules bind and exert action

## Allosteric Binding Site

Ligand molecules bind at one place and action takes place at another site through relay of conformational changes

## Cryptic Binding Site

Ligand molecules bind at one place and action takes place at another site through relay of conformational changes

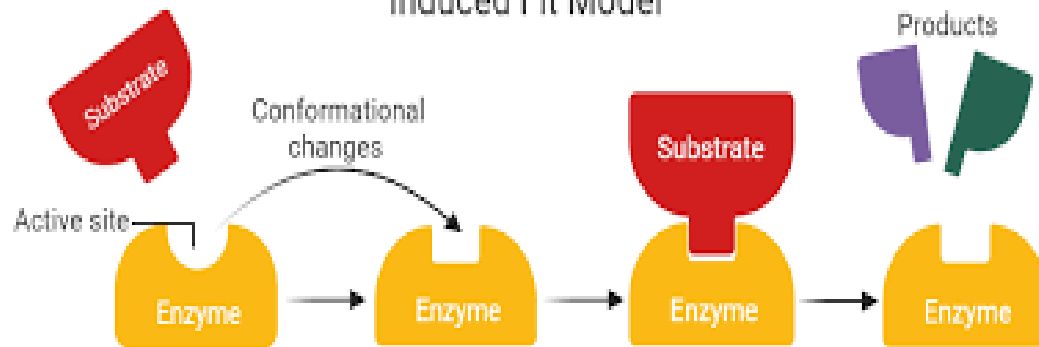
# Binding sites

- Are very important when studying the MD
- Properties
  - Usually are pockets or deep cavities on protein surface
  - Lined by a set of a.a. with different properties (charged, polar, hydrophobic)
  - Complementary to their substrate
  - Is flexible & can accommodate variety of related structures

# Enzyme Action Models

- Lock and key (Rigid shape complementarity)
- Induced fit model (Receptor flexibility)
- Transition State Model  
(Substrate shape is changed in intermediated shape)

Induced Fit Model



Hexokinase without glucose (open)



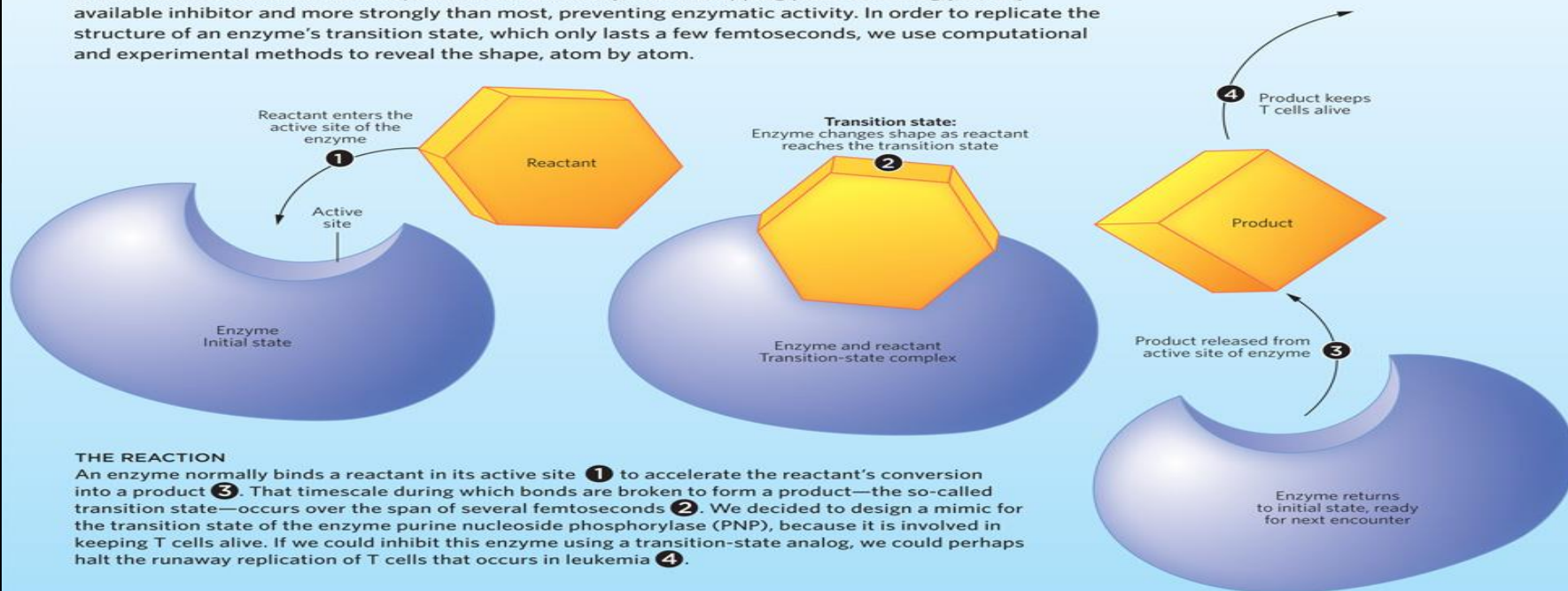
Hexokinase with glucose (closed)



<http://pdb101.rcsb.org/motm/50>

## DESIGNING TRANSITION-STATE INHIBITORS

A transition-state mimic has the power to bind an enzyme at its tipping point as strongly as any available inhibitor and more strongly than most, preventing enzymatic activity. In order to replicate the structure of an enzyme's transition state, which only lasts a few femtoseconds, we use computational and experimental methods to reveal the shape, atom by atom.

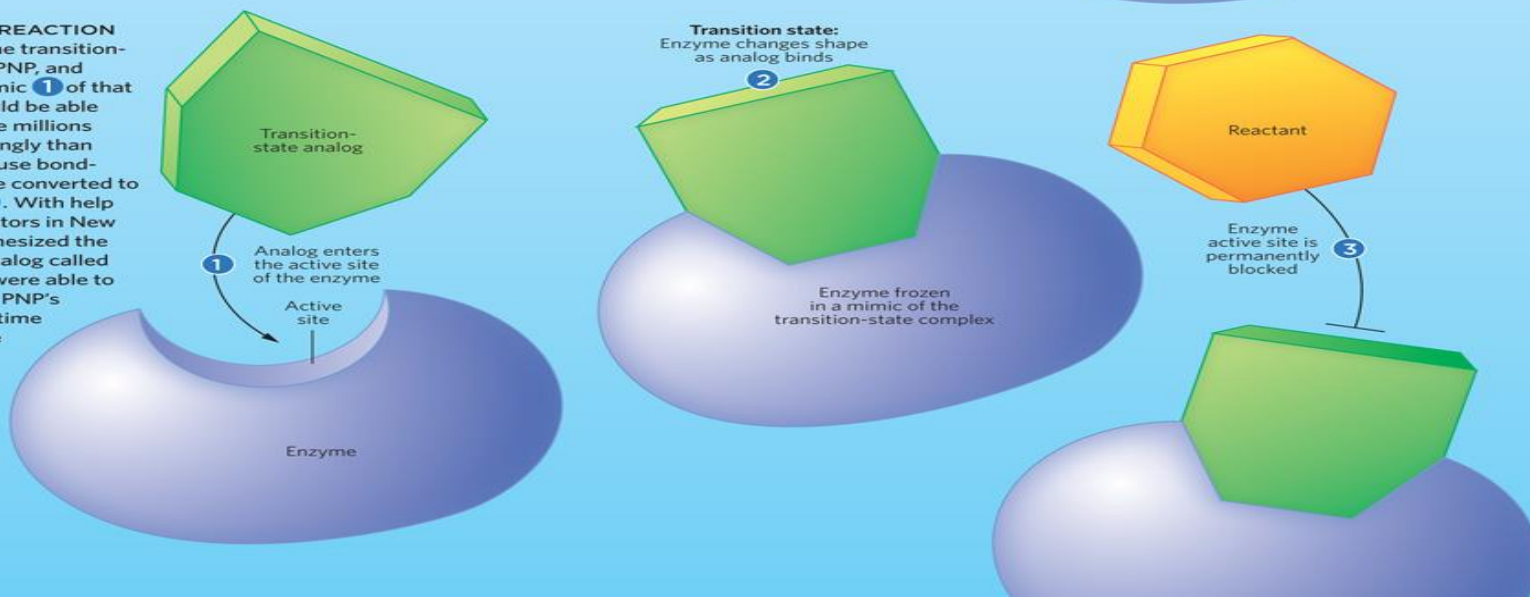


### THE REACTION

An enzyme normally binds a reactant in its active site **1** to accelerate the reactant's conversion into a product **3**. That timescale during which bonds are broken to form a product—the so-called transition state—occurs over the span of several femtoseconds **2**. We decided to design a mimic for the transition state of the enzyme purine nucleoside phosphorylase (PNP), because it is involved in keeping T cells alive. If we could inhibit this enzyme using a transition-state analog, we could perhaps halt the runaway replication of T cells that occurs in leukemia **4**.

### INHIBITING THE REACTION

We first defined the transition-state structure of PNP, and then created a mimic **1** of that structure that would be able to bind the enzyme millions of times more strongly than the reactant because bond-breaking forces are converted to binding energy **2**. With help from our collaborators in New Zealand who synthesized the transition-state analog called immucillin-H, we were able to successfully block PNP's activity for the lifetime of the cell **3**. The transition-state mimic is now in clinical trials for the treatment of several forms of leukemia.

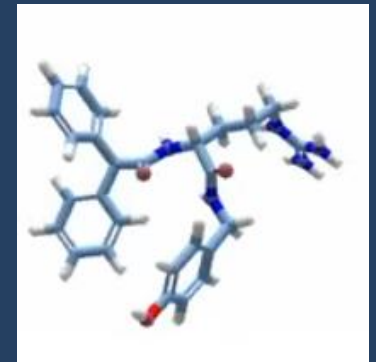
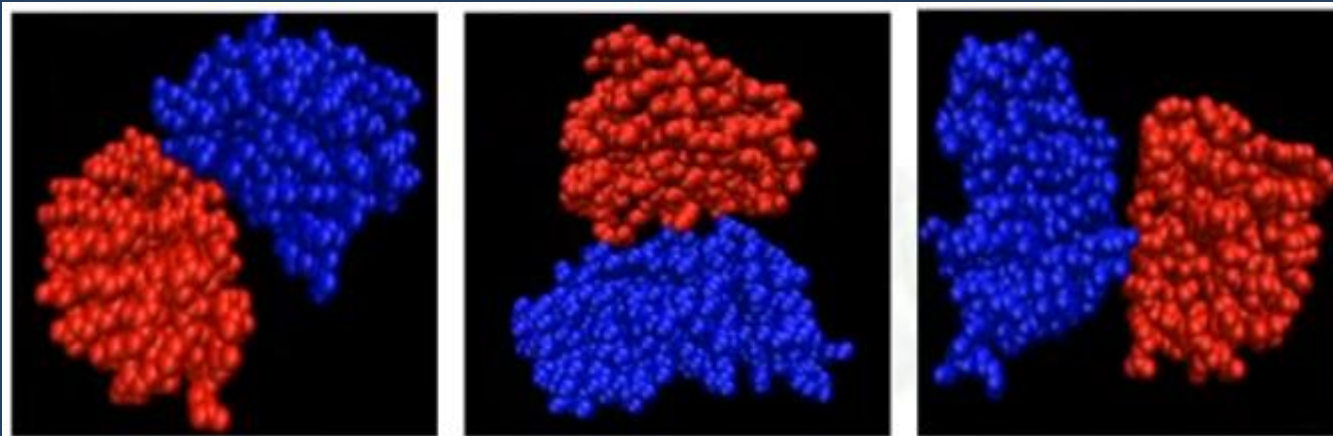


# Molecular Docking

- A computer based technique to investigate molecular interaction
- Attempts to find the most probable binding conformation of 2 molecules which minimize the energy through maximized interaction
- An extension of molecular modeling approaches
- Goal is to find stable complex of receptor and ligand with possible energy

# Molecular flexibility

- Molecules are flexible in nature and many alter each other's structure when interacting
- Thus interacting molecules possess large DOFs (Degree of Freedom)
- Possible conformations are enormous in number





**Thank you  
for listening**