

Managing Soil Salinity and Sodicity: A Biological Perspective

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Presentation to the University of Baghdad
23 March 2024

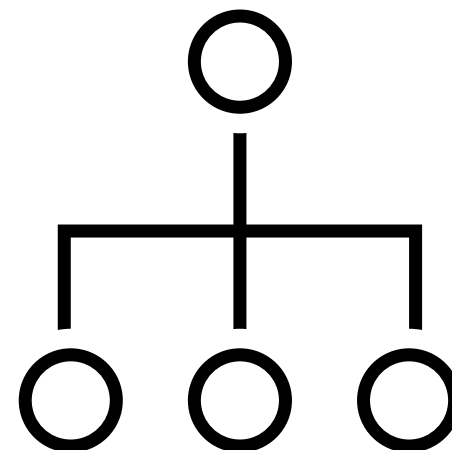


We Do Not Have Salinity and Sodicity Problems in Kentucky

Human Error
Human Activity - Oil Drilling
Wastewater
Natural Events - Saline Seeps

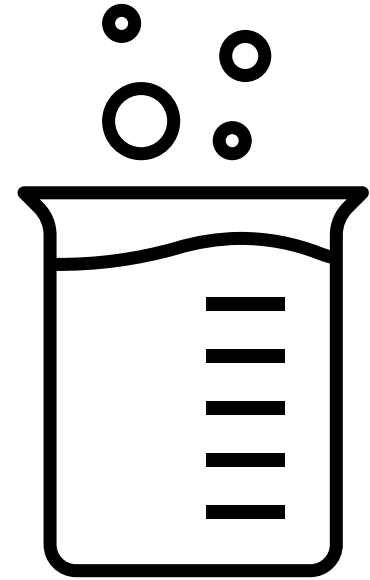
Outline of Presentation

- Review (things you already know)
 - A. What are saline soils?
 - B. What are sodic soils?
 - C. Where and why do they occur?
 - D. Why are they bad for plants and soil biology?
 - E. What are the mitigation strategies?
- How do you manage the biological environment?
 - A. Structure
 - B. Microbes

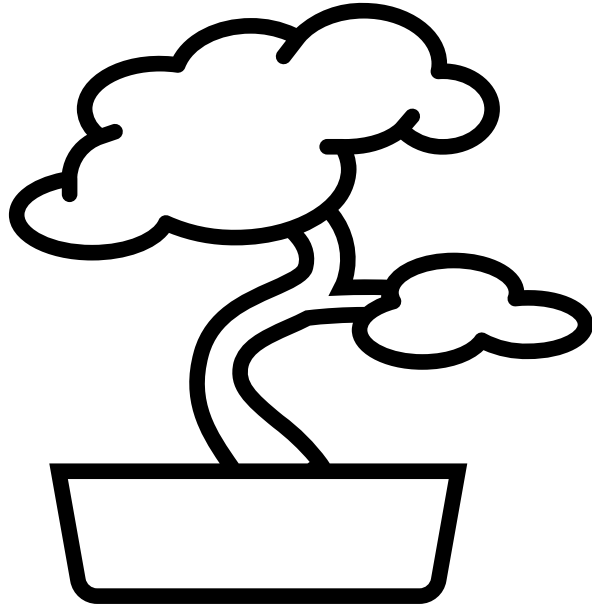


Chemical Degradation

- Salinization
- Sodification
- Alkalinization
- Acidification
- Nutrient Depletion
- Nutrient Excess
- Agrichemicals
- Industrial Contamination



Directly Affects Plant Growth



Indirectly Affects Growth by
Influencing Soil Biological and
Physical Properties

Acidity/Alkalinity

ultra acid	< 3.5
extremely acid	3.5 to 4.4
very strongly acid	4.5 to 5.0
strongly acid	5.1 to 5.5
moderately acid	5.6 to 6.0
slightly acid	6.1 to 6.5
Neutral	6.6 to 7.3
slightly alkaline	7.4 to 7.8
moderately alkaline	7.9 to 8.4
strongly alkaline	8.5 to 9.0
very strongly alkaline	> 9.1

Affects Nutrient Availability (P/Fe), Nutrient Toxicity (Al/Mn/Mo), Plant Growth

Individual Microbes Have pH Minimum and Maximum



Salt- and Sodium-Affected Soils Are in:

- Arid regions
- Irrigated regions with high water tables
- Irrigated regions with poor water quality
- Saline seeps after fallow

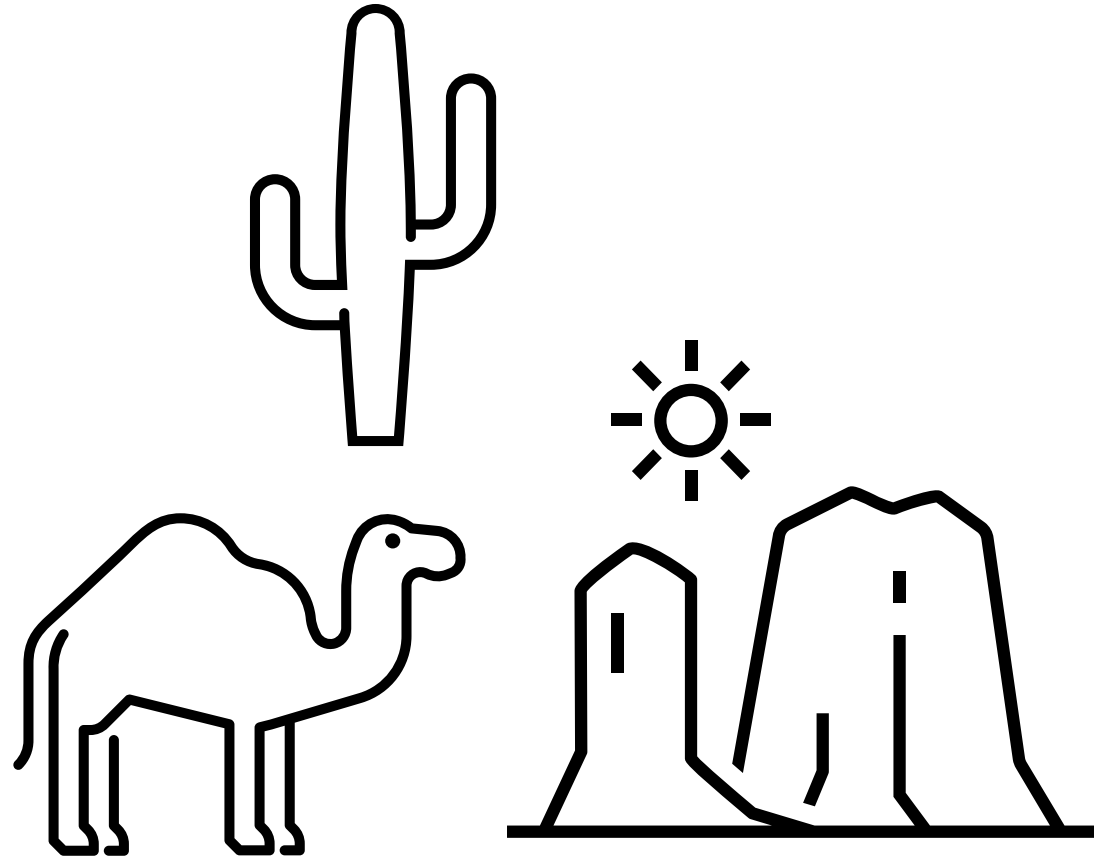


Table 2. Salt classifications in soils (NRCS).

Characterization	EC†	ESP‡	SAR _e §	pH#
Normal	<4	<15	<12	<8.5
Saline	>4	<15	<12	<8.5
Sodic	<4	>15	>12	>8.5
Saline-Sodic	>4	>15	>12	>8.5

† EC, electrical conductivity, dS m⁻¹.

‡ ESP, exchangeable sodium percentage.

§ SAR_e, sodium adsorption ratio determined from a saturated paste extract.

pH, term describing the concentration of hydrogen.

General Characteristics

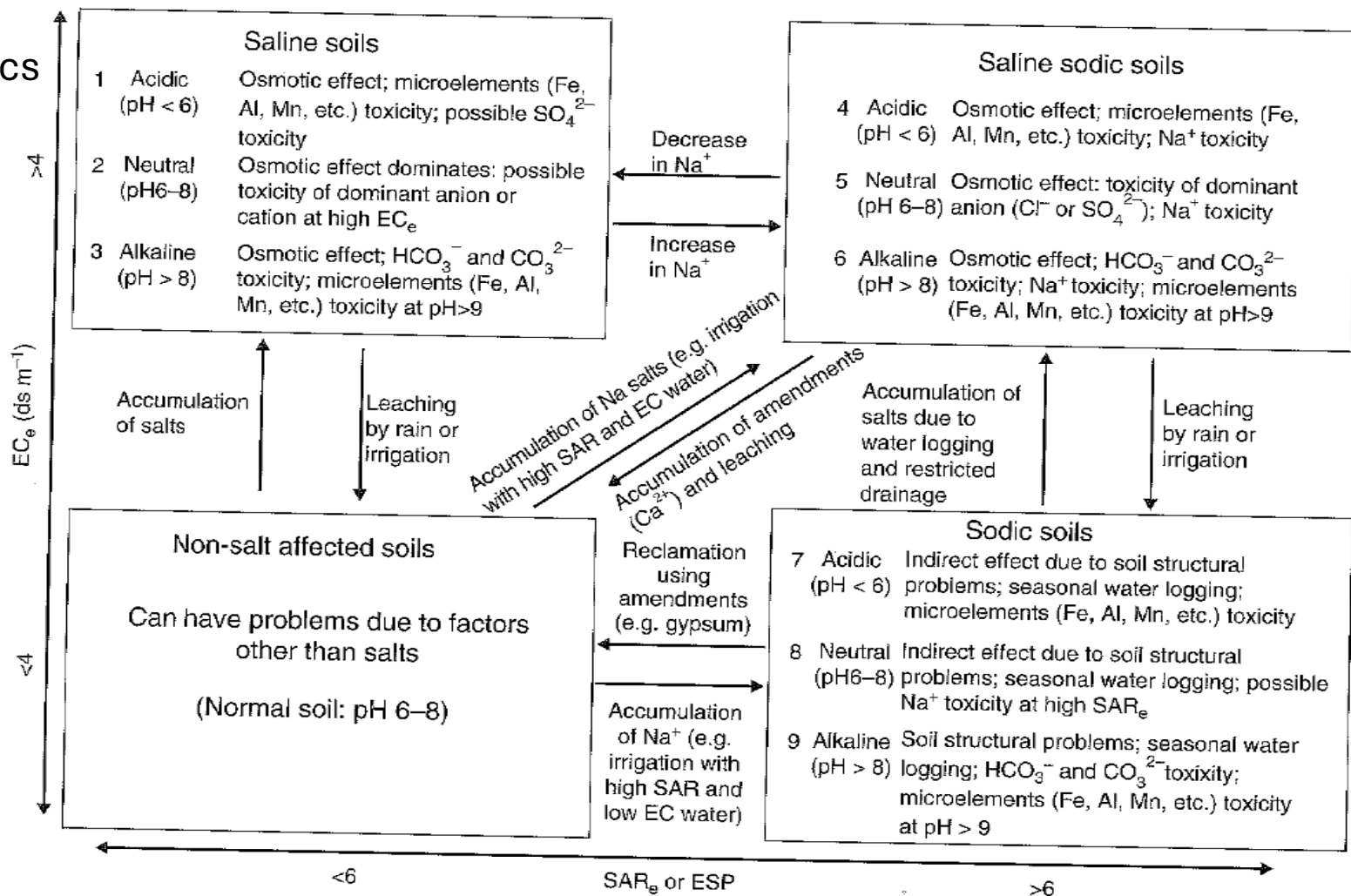


Figure 7.7 Key properties of the major types of salt-affected soils and possible mechanisms of impact on plant growth. Soils are categorised absorption ratio (SAR_e) and electrical conductivity (EC_e) measured in soil saturation extracts and pH 1:5 measured in soil:water suspension. The (SAR_e ≈ ESP) of 6 and above to classify a soil as sodic is based on the Australian soil classification criteria (Isbell, 1996); however, the general are of relevance worldwide. Toxicity, deficiency or ion-imbalance due to other elements (e.g. B, K, N, P) will depend on the ionic composition of and the particular species of plant.

Salinity – Too Much Salt!

3/26/2024

Salinity – EC

Measured by Electrical Conductivity (EC, dS m^{-1})

EC = electrical conductance of a saturated paste solution

Salts with solubility greater than that of gypsum (a less soluble salt)

Salic (saline) soil: $\text{EC} > 30 \text{ dS m}^{-1}$

The upper EC limit for drinking water is 1.6 dS m^{-1} ($1 \text{ dS m}^{-1} = 1 \text{ mmho cm}^{-1}$)

Table 1. Common soluble salts found in soils (FAO, 1973).

Compound	Common name	Molecular formula	Solubility (20 °C)
Calcium carbonate	lime	CaCO ₃	0.06†
Magnesium carbonate		MgCO ₃	2.51‡
Sodium carbonate	soda	Na ₂ CO ₃	179
Sodium bicarbonate	baking soda	NaHCO ₃	87
Calcium sulfate	gypsum	CaSO ₄	1.9
Magnesium sulfate	epsom	MgSO ₄	252
Sodium sulfate	glauber's salt	Na ₂ SO ₄	161
Magnesium chloride		MgCl ₂	410
Sodium chloride	table salt	NaCl	264
Calcium chloride		CaCl ₂	427

† Solubility of CaCO₃ in 0.00032 atm of CO₂.

‡ Solubility of MgCO₃ in 0.005 atm of CO₂ at 18°C.

Table 3. Salinity classes of soils (NRCS).

Classes	Electrical conductivity
	dS m ⁻¹
Nonsaline	0-2
Very slightly saline	≥2-4
Slightly saline	≥4-8
Moderately saline	≥8-16
Strongly saline	≥16

Salinity and Plant Damage

- Due to the lower osmotic potential in saline soils, plants commonly exhibit **water stress symptoms**: wilting, stunting with cupped leaves, browning and brittle leaf tips and margins, and potentially become deep blue-green from excessive wax accumulation.
- Reduced P, K, or essential ions due to high Ca concentrations.
- Reduced K and Ca uptake because of high Na concentrations.
- Reduced nitrate-nitrogen uptake due to excess sulfate and chloride.
- The toxic effects of excess levels of Na and Cl on plants include leaf-tip burn, scorching of the leaf margin, chlorosis, and premature leaf drop.

Wilting Occurs Sooner With Higher Salt (EC) Content in Soil Water – Salt Decreases Water Potential (ψ)

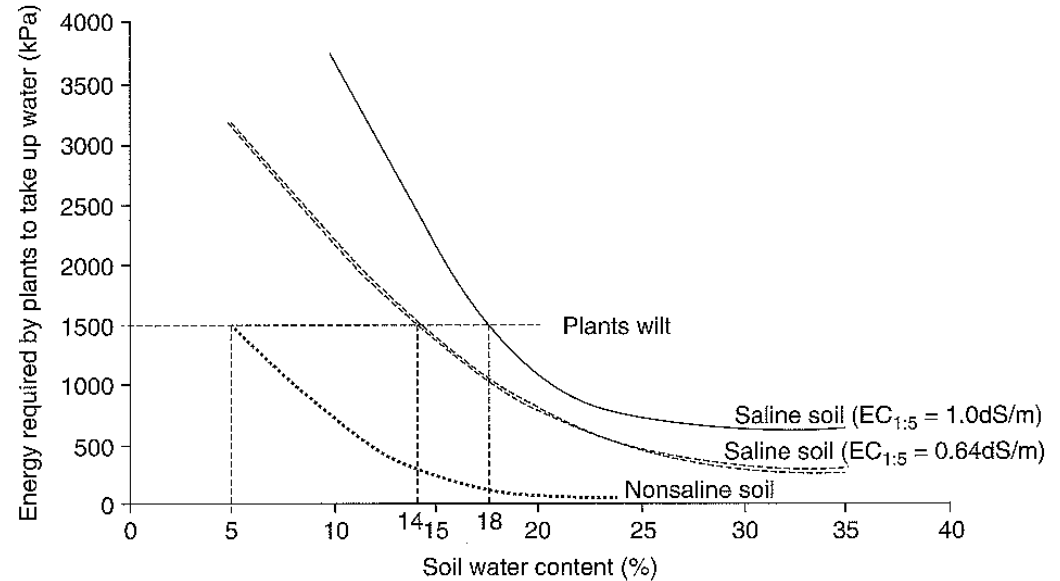


Figure 7.8 Energy (equivalent to soil matric plus osmotic potential) required by plants to take up water from a loamy soil as influenced by EC_{1:5} and % soil-water content.

Note: 1500 kPa = 1.5 Mpa = 15 Atm

Soil Conditions and Plant Growth, Edited by Peter J. Gregory and Stephen Nortcliff.
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Soil Microbes and Salinity

Microbes vary from being very sensitive to low osmotic potential to very resistant

Gram negative bacteria are sensitive

Fungi and Gram positive bacteria are more resistant

Halophiles grow in environments that are incredibly salty

Increasing salinity decreases the diversity of microbes in the soil and selects for resistant organisms

Solutions to Salinity



IMPROVE DRAINAGE



IRRIGATE WITH GOOD
WATER



PLANT PLACEMENT



PLANT SELECTION
- SALT TOLERANT
- SALT RESISTANT



SALT TOLERANT
SYMBIONTS
- MYCORRHIZAE
- PGPR



ORGANIC MATTER TO
REPLACE SOURCES OF
SALT AS FERTILIZER

What to Do?

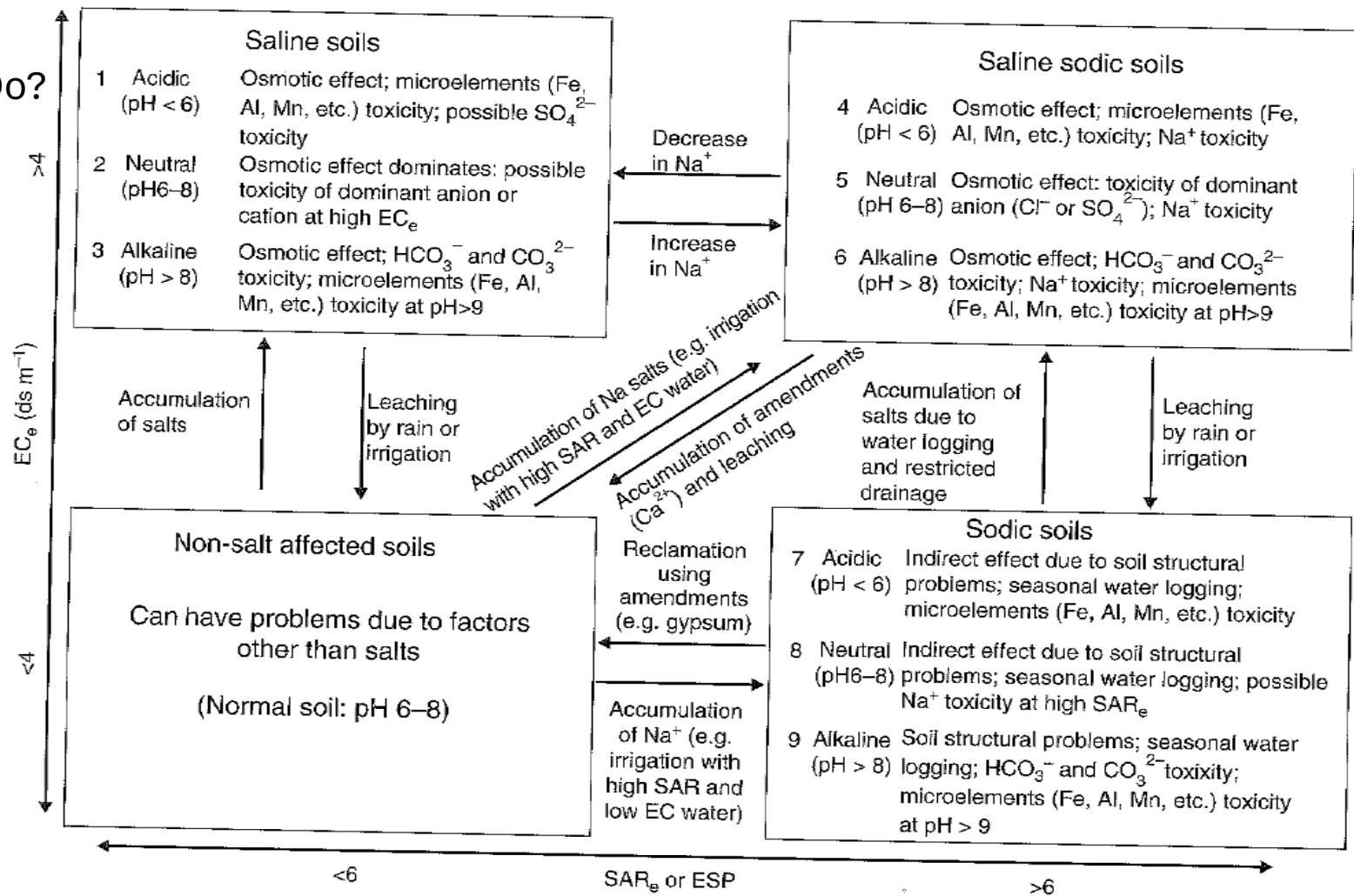


Figure 7.7 Key properties of the major types of salt-affected soils and possible mechanisms of impact on plant growth. Soils are categorised by absorption ratio (SAR_e) and electrical conductivity (EC_e) measured in soil saturation extracts and pH 1:5 measured in soil:water suspension. The (SAR_e ≈ ESP) of 6 and above to classify a soil as sodic is based on the Australian soil classification criteria (Isbell, 1996); however, the general are of relevance worldwide. Toxicity, deficiency or ion-imbalance due to other elements (e.g. B, K, N, P) will depend on the ionic composition of and the particular species of plant.

What is a Sodic Soil and Why Does it Matter?

- Sodic soils have excess Na on the soil's exchange sites. The dominant cations on the soil exchange sites are typically Ca^{+2} , Mg^{+2} , and K^{+} . As the concentration of Na^{+} increases in soil, Na^{+} replaces other cations on the soil exchange sites by mass action.
- The result can be slaking, swelling, and clay dispersion of soil particles upon wetting.
- **Slaking** is the breakdown of soil aggregates into particles >20 μm in size.
- **Dispersion** is the breakdown of fine soil particles to <2 μm in size.
- The NRCS classifies soils as sodic if the SAR_e is >12 , the ESP is >15 , the EC is <4.0 dS m^{-1} , and the pH is >8.5

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Sodicity – SAR & ESP

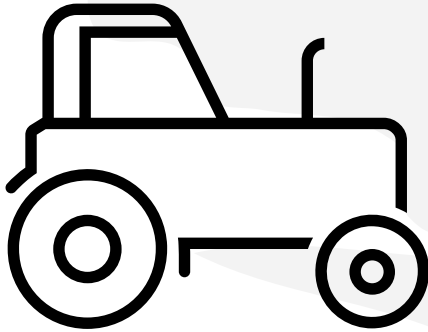
- SAR = sodium adsorption ratio = $\text{Na} / \sqrt{(1/2[\text{Ca} + \text{Mg}])}$

all expressed in meq/L, in a saturated paste extract

- ESP = exchangeable sodium percentage = $100(\text{Na}_{\text{ex}} / \text{CEC})$
all in meq/100 g soil.

	SAR
slight	< 13
moderate	13-30
strong	> 30

Sodicity Damage



- Plants experience nutrient imbalance and water stress similar to that with salinity
- Plants suffer from soil physical changes
- When soils with excess Na are drained or washed with high quality irrigation water they can get dispersion, loss of structure, pore plugging with dispersed clay, low percolation and infiltration; crusting; hard-setting

Dispersion



- Clay particles become repulsive to one another.
- Structure breaks down - large pores collapse
- Clay particles move downward, plug small pores
- The opposite of dispersion is flocculation.
- What is behind the dispersion problem with sodic soils?
 - Large sphere of hydration for Na

Salinity and Sodictity

- Salinity and sodicity exasperate water stress
- Drought strategies
 - Controlled irrigation: furrow, drip, sprinkler
 - Resistant resources: phenotypes and breeding
 - Agronomic: residue, rotation, and position
 - Bioresources: PGPR



What to Do?

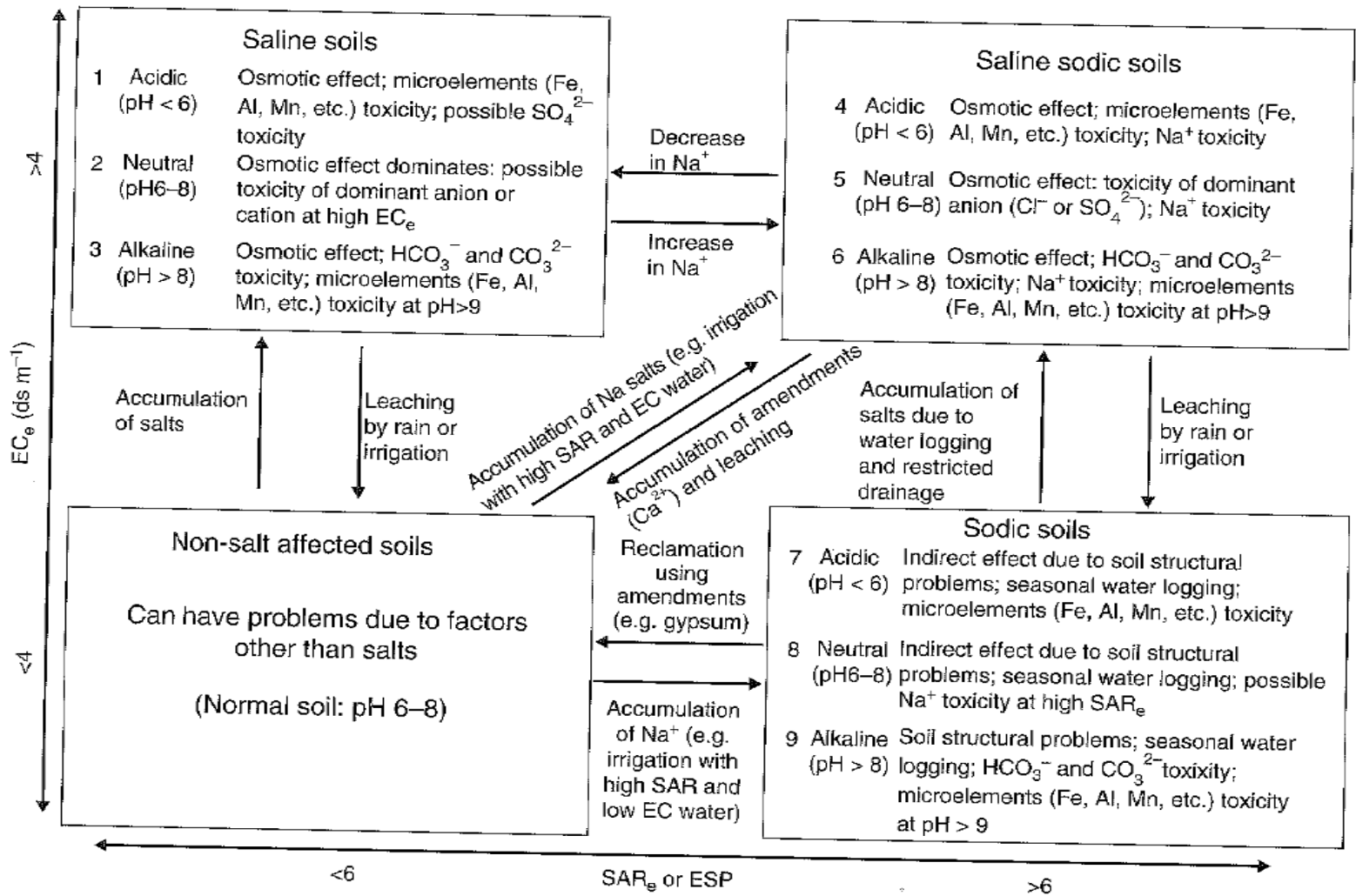
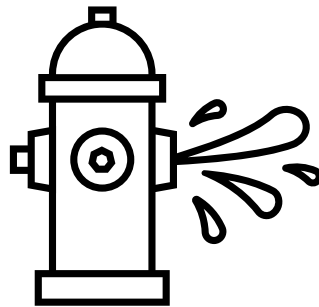


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Removing Excess Salinity


- You need water, water above and beyond what is needed to grow the crop
- The salt “Leaching Requirement”
- $LR = EC_w / [5(EC_e) - EC_w]$ where EC_w is the electrical conductance of the irrigation water and EC_e is the electrical conductance of the soil at which no yield loss occurs. The EC_e value differs with the chosen crop.
- The fraction of additional water needed.



Managing Sodictity

- You need to remove Na, without causing dispersion – you need both water AND Ca
- Why can't you use Ca carbonate?
- Gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$) is a cheap and more soluble Ca source
- The gypsum requirement, GR:

$$\text{GR} = 0.0086\text{FD}_s\rho_d\text{CEC}(\text{ESP}_i - \text{ESP}_f).$$



The Biological Perspective for Chemically Degraded Soils



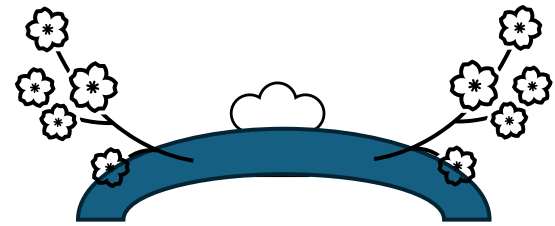


Acidity and Alkalinity

Raise pH if it is acid
- Lime with CaCO_3 or a
Mg/Ca oxide

Lower if the pH is
alkaline
- Everyone mineralizes,
few solubilize when
alkalinity is an issue

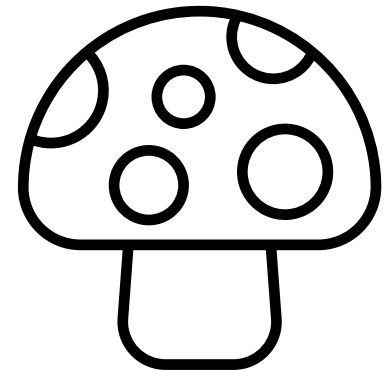
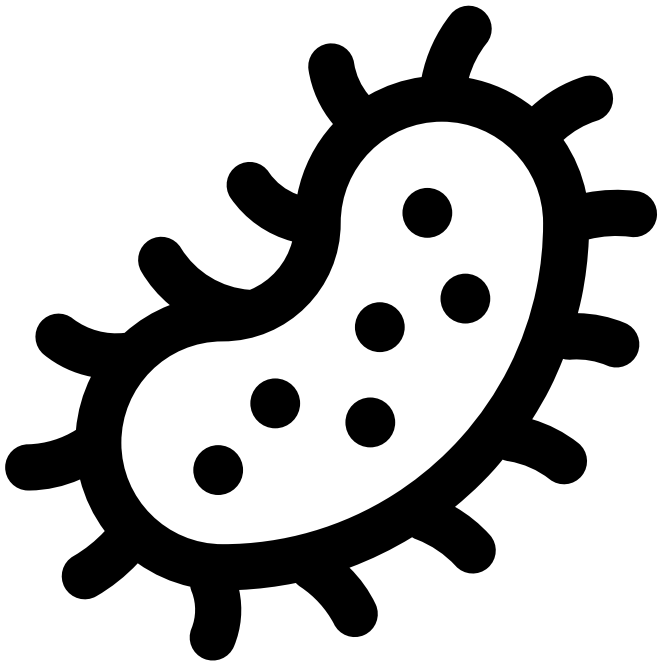
Agronomic Solutions



- Ridge planting avoids the most salt-affected soil - a desperate solution.
- Increasing soil residue will decrease evapotranspiration and keep soil cooler
 - Maintains more water in soil and reduces plant stress
- Increasing soil organic matter improves soil structure and release soil nutrients
 - Reduces dispersion and increases infiltration
 - Helps to minimize soil acidity effects by increasing CEC
 - Mineralization releases nutrients that replace chemical salts
 - Unless the organic matter is high in salts (e.g. poultry litter)
- Legumes replace chemical N as the N source and preserve available P

Microbial Solutions

Find Microbes That Persist in Saline or Sodic Environments and Improve Plant Growth in Water Stressed Environments





Pathways for Drought Tolerance Through Rhizosphere Bacteria

- Create phytohormones
- Produce the enzyme ACC-deaminase, which lowers the amount of ethylene
- Cause induced systemic resistance
- Develop biofilms
- Produce extracellular polysaccharides



Stress Tolerant Microbes That Enhance Growth

- Auxins, cytokinins, and gibberellins are among the phytohormones produced by rhizobacteria that help plants grow.
- Crop plants exposed to beneficial bacteria develop extensive roots, and root hairs, all of which may enhance the plant's capacity to endure various stressful environments
- P solubilizing bacteria reduce the need for chemical P fertilization

Microbes That Reduce Stress Hormones

- Ethylene is a plant stress hormone.
- The metabolic pathway leading from from methionine to ethylene is:

Methionine →
S-adenosylmethionine →
1-aminocyclopropane-1-
carboxylic acid (ACC) →
ethylene (C₂H₄)

- Microbes with ACC deaminase reduce ethylene production and promote plant growth during water stress.



Common Genera Bacteria That Can be PGPR

- *Acetobacter*
- *Azospirillum*
- *Bacillus* (e.g. *B. subtilis*)
- *Burkholderia*
- *Pseudomonas*
- *Rhizobium*

PGPR = Plant Growth Promoting Rhizobacteria





Selection Strategies

- + ● • Pick isolates that are inherently resistant to high salt and Na and water stress
- • Increase mass in environments that mimic the stressed conditions
- Inoculate in high numbers early
- One interesting approach is to use rhizobia with PGPR properties and rotate with legumes
 - this maintains the population of the PGPR in the soil environment with time.
 - Rotation is good anyway



How Do Mycorrhizae (AMF) Improve Salt Tolerance?

- Enhance nutrient acquisition (P, N, Mg and Ca)
- Maintain the $K^+ : Na^+$ ratio
- Accumulate non salt osmolytes: proline, betaines, polyamines, carbohydrates and antioxidants
- Promote physiological changes (photosynthetic efficiency, relative permeability, water status, abscissic acid accumulation, nodulation and nitrogen fixation)
- Induce molecular changes (the expression of genes: *PIP*, Na^+/H^+ antiporters, *Lsnced*, *Lslea* and *LsP5CS*)
- Stimulate ultra-structural changes.

Mycorrhizae

Increase

Increase the volume of soil exploited for nutrients and water

Have

Have the capacity to take up water at lower potential than plant roots

Develop

Develop improved aggregation in the soil surrounding roots, which increases percolation

Extend

Extend the lifespan of roots for water sorption

Mechanisms through which AMF reduce impacts of high salinity on plants

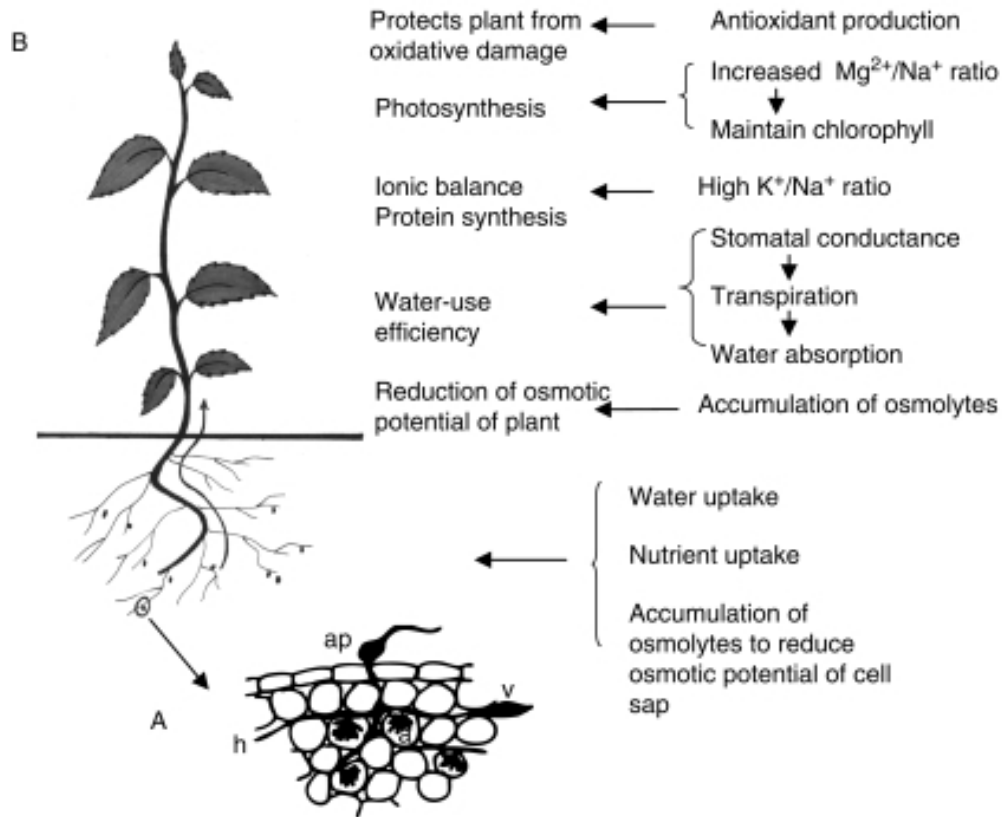



Fig. 2

The intricate functioning of arbuscular mycorrhizal (AM) fungi in ameliorating salt stress in plants. In AM symbiosis, the fungus forms an appressorium (ap) on the root surface and enters the root cortex by extending its hyphae (h). The hyphae form arbuscules (a) and vesicles (v) in the cortex. Salinity deprives plants of the basic requirements of water and nutrients, causing physiological drought and a decrease in osmotic potential accompanied by nutrient deficiency, rendering plants weak and unproductive. Arbuscular mycorrhiza help plants in salt stress by improving water and nutrient uptake: a decrease in osmotic potential is countered by increasing accumulation of osmolytes, and water-use efficiency, photosynthesis and antioxidant production (to scavenge ROS) is more efficient in salt-stressed plants in the presence of AMF (see text).



Salinity Reduces Spore Formation

Reduces Overall
Mycorrhizal Population and
Diversity

Mycorrhizae still colonize
roots at very high salinity

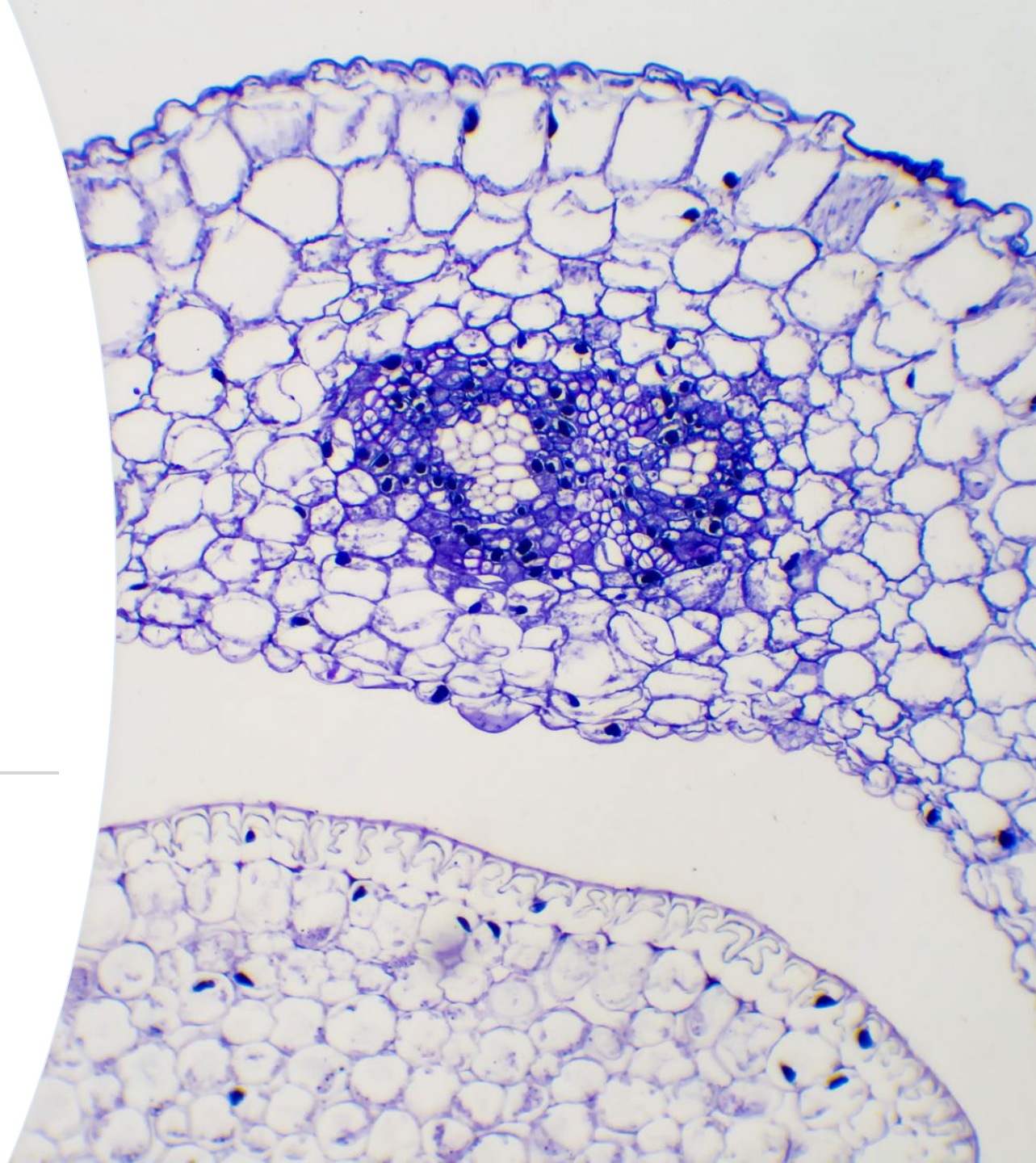


Table 1. Studies on percentage of AMF colonization in root under salinity stress

* The range of salinity within brackets is the actual salt concentrations used by the authors.

Range of salinity*	Plant	Fungus	% root colonization in AM plants	References
1.4–7.4 dS m ⁻¹	<i>Lycopersicon esculentum</i>	<i>Glomus mosseae</i>	49.6–36	Al-Karaki (2000)
2–12 dS m ⁻¹	<i>Lactuca sativa</i>	Mixture of <i>Glomus</i> , <i>Acaulospora</i> and <i>Entrophora</i> spp. procured from (a) saline playa or (b) non-saline vegetable farm	(a) 43.0–26.2 (b) 34.8–29.9	Cantrell and Linderman (2001)
2–12 dS m ⁻¹	<i>Allium cepa</i>	Mixture of <i>Glomus</i> , <i>Acaulospora</i> and <i>Entrophora</i> spp. procured from (a) saline playa or (b) non-saline vegetable farm	(a) 61.7–38.8 (b) 28.8–18.0	Cantrell and Linderman (2001)
0–13.19 dS m ⁻¹ (0–100 mm)	<i>Zea mays</i>	<i>Glomus mosseae</i>	70–80	Feng et al. (2002)
0–6.10 dS m ⁻¹ (0–3 g kg ⁻¹)	<i>Gossypium arboreum</i>	<i>Glomus mosseae</i> : (a) GM1 from non-saline soils; (b) GM2 from saline soils	(a) 38 ± 3 to 15 ± 2 (b) 46 ± 5 to 21 ± 3	Tian et al. (2004)
40.2 dS m ⁻¹	<i>Tamarix chinensis</i> , <i>Phragmites communis</i> , <i>Suaeda glauca</i> , <i>Aeluropus littoralis</i> var. <i>sinensis</i> and <i>Cirsium setosum</i> (in Yellow River Delta, China)	Mixture of <i>Archaeospora</i> , <i>Acaulospora</i> and <i>Glomus</i>	0.2–9.5	F. Y. Wang et al. (2004)

Sources

- [Evelin, H., R. Kapoor, and B, Giri. 2009.](#)
Arbuscular mycorrhizal fungi in alleviation of salt stress: a review. [Ann Bot.](#) 2009. 104(7): 1263–1280.
- Tahir, M.N. 2024. Evaluating the effect of microbial consortium for improving growth and quality of sorghum and residual effect of inoculation on berseem under water deficit conditions. PhD dissertation. University of Faisalabad, Pakistan.



Questions?